

Mass-spectral Rearrangement of Stannonium Ions: Migration of Oxygen from Carbon to Tin and Elimination of C_6H_4 in Some Phenoxastannins†

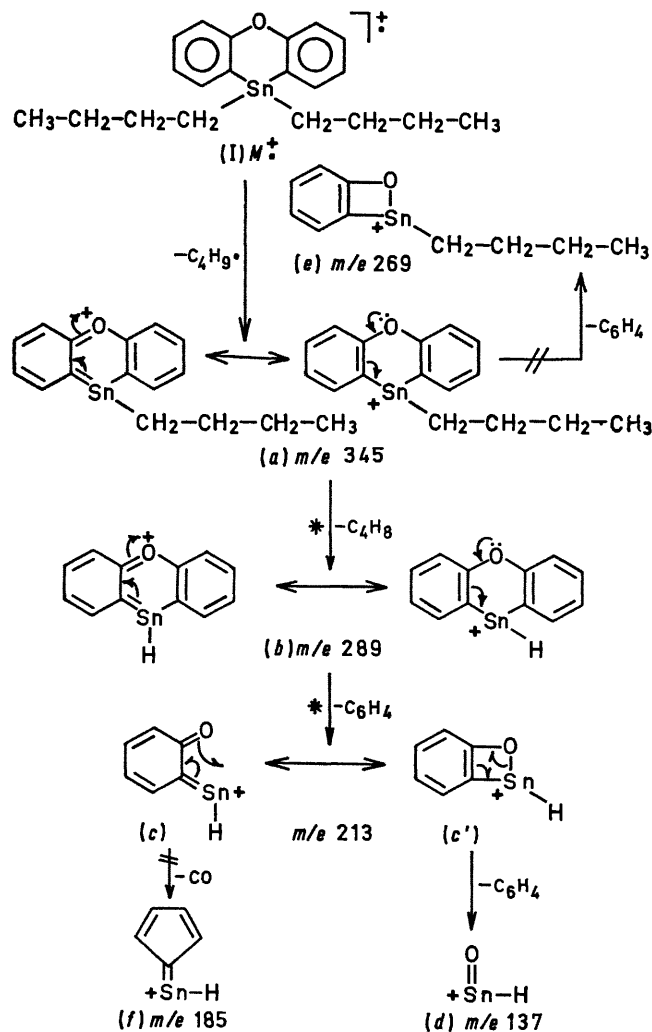
By ISTVÁN LENGYEL* and MICHAEL J. AARONSON

(Department of Chemistry, St. John's University, Jamaica, New York 11432)

Summary The mass spectra of some 10,10-dialkyl phenoxastannins reveal a novel elimination-rearrangement reaction involving expulsion of a C_6H_4 fragment and Sn-O bond formation.

EVEN-ELECTRON fragment ions in the mass spectrometer frequently undergo elimination of a neutral fragment to form a new even-electron ion of lower mass number.¹ The neutral fragment expelled is usually a small organic or inorganic

molecule of high stability, such as C_2H_2 , C_2H_4 , CH_2O , CH_2CO , CO , CO_2 , HCN , $HNCO$, H_2O , etc., and its stability often supplies an important part of the driving force for fragmentations of this type.



We now report a novel electron-impact induced reaction involving the migration of oxygen from carbon to tin and elimination of the C_6H_4 (benzyne?) fragment, observed in

the mass spectra of some 10,10-dialkyl-substituted phenoxastannins. There is no previous published report on the mass spectra of these compounds.†

The major primary step in the electron-impact induced fragmentation of 10,10-di-n-butylphenoxastannin² (I; m.p. 118—119°) at 50—70 eV[‡] is the ejection of a C_4H_9 radical to give the resonance-stabilized even-electron ion (a), m/e 345,§ $C_{16}H_{17}OSn$,¶ which in turn loses the elements of butene with concomitant rearrangement of a hydrogen atom from the alkyl ligand to the metal, leading to the resonance-stabilized even-electron ion (b), m/e 289, $C_{12}H_9OSn$. It is noteworthy that the energy requirement of the alternative elimination of C_6H_4 , that would lead to ion (e), is sufficiently high to suppress the latter process completely. Both ions (a) and (b) are major species in the spectra even at 12 eV. One of the pathways of the further fragmentation of ion (b) involves elimination of C_6H_4 to afford the resonance-stabilized ion [c(c'),] m/e 213, C_6H_5OSn . The formation of ion [c(c')] requires fission of several bonds and the loss of the resonance energy of the aromatic ring in the neutral particle: its presence strongly suggests anchimeric assistance by the neighbouring oxygen function resulting in O—Sn bond formation.** That ion m/e 213 is probably more accurately represented as (c') rather than (c), can be surmised from the absence of ion (f): the quinoid structure (c) would be expected to eliminate CO.

Ion [c(c')] subsequently eliminates a second C_6H_4 particle yielding ion (d), m/e 137, $HSnO$. Both ions [c(c')] and (d) are absent in 12 eV scans. Entirely analogous processes have been observed in the mass spectrum of 10,10-diethylphenoxastannin.

The occurrence of this remarkable elimination–rearrangement reaction may be attributed to a combination of several factors: (i) the pronounced tendency of tin to form further even-electron ions⁵ in which the tin is in the Sn^{IV} oxidation state; (ii) the favourable arrangement of atoms for a four-membered transition state such that the ring can regain aromaticity by bond formation between tin and oxygen; (iii) the stability of the ionic product ($HSnO$) which contains the strong $Sn=O$ double bond. A number of other interesting rearrangements were also revealed by the mass spectra of 10,10-disubstituted phenoxastannins. Details of these will be published shortly elsewhere.

We thank Professor E. J. Kupchik, Department of Chemistry, St. John's University, for kindly providing the samples used in this investigation, and Professor K. Biemann and his associates, Department of Chemistry, Massachusetts Institute of Technology, for the element maps.

(Received, December 1st, 1969; Com. 1828.)

† A brief report on some aspects of the mass spectra of phenoxastannins was given by I. Lengyel, M. J. Aaronson, and J. P. Dillon at the 17th Annual Conference on Mass Spectrometry and Allied Topics, May 15—23, 1969, Dallas, Texas; cf. Abstracts of Papers, p. 403.

‡ Conventional ("low resolution") mass spectra were recorded on both a CEC 21-103C and a Hitachi RMU-6D instrument, with introduction of the sample *via* a direct probe.

§ All masses are given for the ^{120}Sn isotope.

¶ The elemental compositions of all ions in the spectrum were determined from high resolution spectra ("element maps") recorded on a CEC 21-110B doubly focusing instrument in conjunction with an IBM 1800 computer system.

** While one example of the electron-impact induced rearrangement of a methoxyl group from carbon to silicon has just been published (ref. 4) no bond formation between oxygen and tin has previously been reported.

¹ F. W. McLafferty, "Interpretation of Mass Spectra," W. A. Benjamin, New York, 1966, ch. 8.

² E. J. Kupchik, J. A. Ursino, and P. R. Boudjouk, *J. Organometallic Chem.*, 1967, **10**, 269.

³ K. Biemann, P. Bommer, and D. M. Desiderio, *Tetrahedron Letters*, 1964, 1725.

⁴ W. P. Weber, R. A. Felix, and A. K. Willard, *J. Amer. Chem. Soc.*, 1969, **91**, 6544.

⁵ D. B. Chambers, F. Glockling, and J. R. C. Light, *Quart. Rev.*, 1968, **22**, 317.